



Device for Quantifying Acute Compartment Syndrome

Client: Dr. Christopher Doro

Advisor: Melissa Skala

Team: Compartment Syndrome

Jahnvi Puranik: Team Leader

Haleigh Simon: Communicator and BSAC

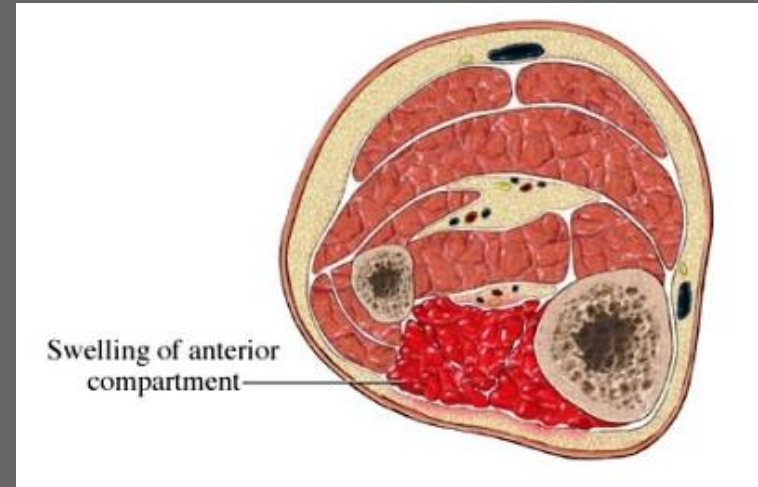
Christian Naxi: BPAG

Jiayi Lin: BWIG



Overview

- Background
- Problem Statement
- Product Design Specifications
- Preliminary Designs
- Design Matrix
- Proposed Final Design
- Future Work



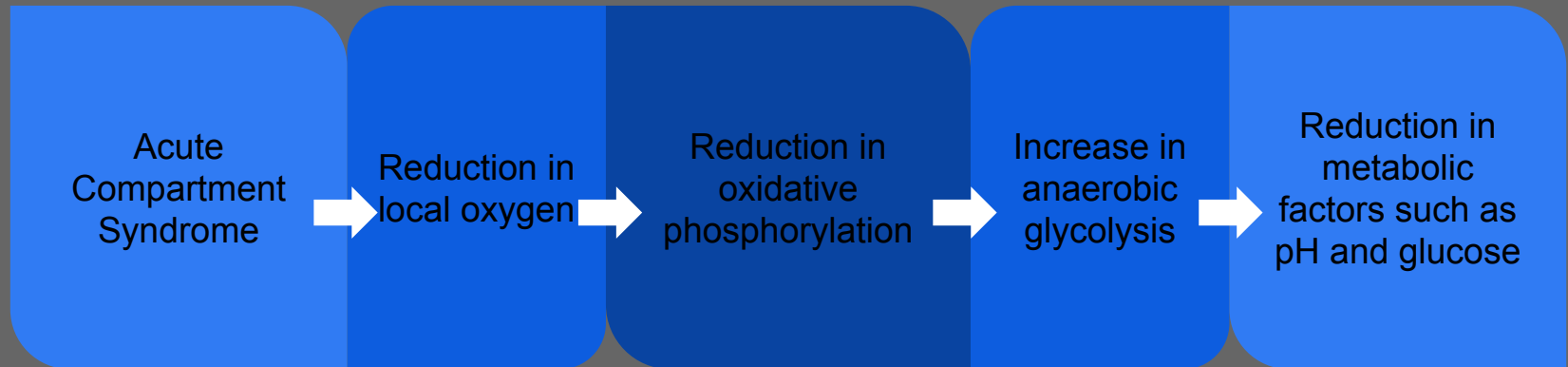


Background

- Increase in pressure in a muscle cavity
- Typically a result of internal bleeding or swelling
- Can lead to muscle death
- Oxygen levels decrease
- Rare in the US, but common in the military



[2]



[3]



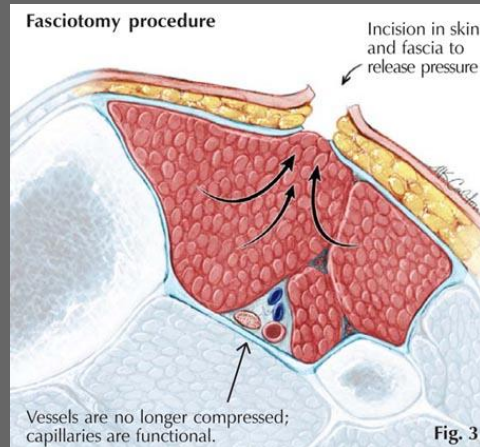
Background- Treatment

Some causes of Acute Compartment Syndrome...

- Crush injuries
- Prolonged limb compression
- Bone fracture wounds
- Gun shot wounds

Emergency Fasciotomy is the standard treatment

1. Decompression
2. Fracture stabilization
3. Vascular repair



[6]



[4]

[7] [5]



Background- Quantifying a Compartment by Pressure

Various studies have published differing results regarding the pressure of an acute compartment

- 30 mm Hg [8]
- 20 mm Hg [9]

Pressure surrounding tibia fractures with no sign of compartment syndrome

- 30 mm Hg diversion from diastolic

Misdiagnosis rate: 35%

Problem Statement



- Develop a device that can quickly, and safely, quantify an existing or developing acute compartment
 - Reduce the false-positive rate of diagnosis
 - Prevent unnecessary surgery
 - Reduce potential muscle damage by decreasing time prior to treatment

Product Design Specifications



Performance requirements:

- Measure metabolite inside a muscle compartment *in vivo*
 - Within 5% accuracy
- Device output should display measurement that is easily readable by the user
- Calibrate time should be less than 5 minutes

Size

- Device must fit in 16 gauge needle (1.2 mm in diameter)
- Probe length of 4-6 cm

Preliminary Designs

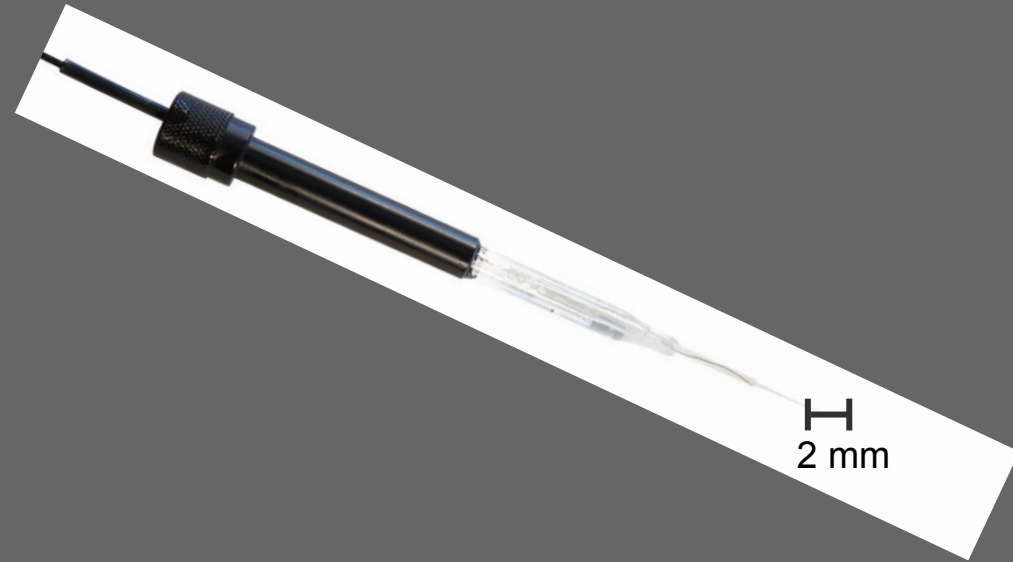


Glucose Probe



[10]

Ion-selective field effect transistor (ISFET) Probe

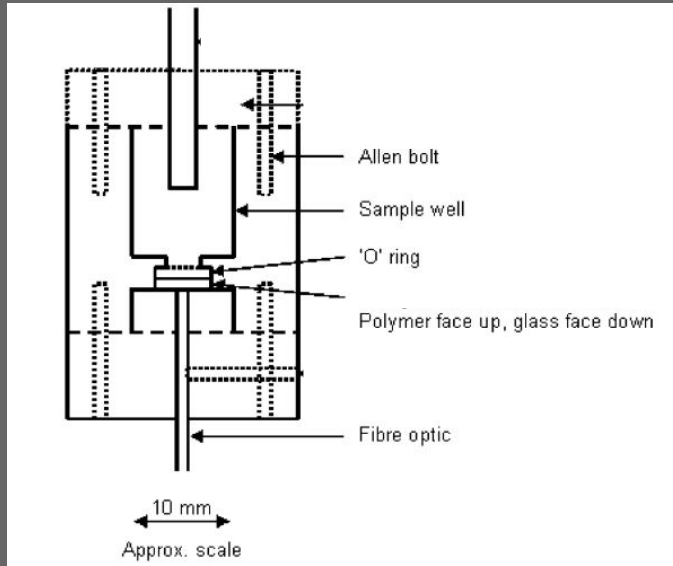


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Preliminary Designs

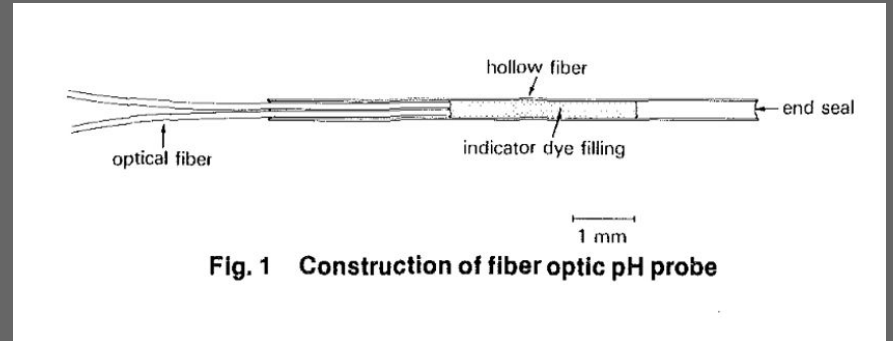


pH polymer probe



[12]

pH optic sensor

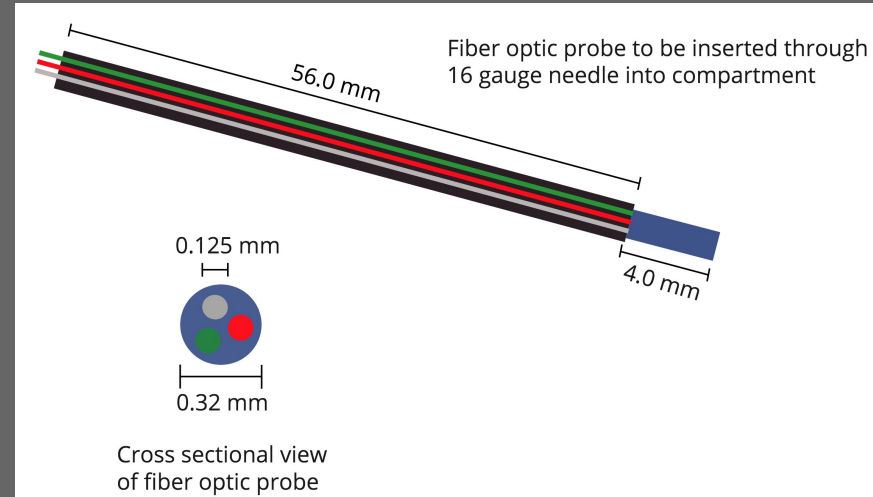
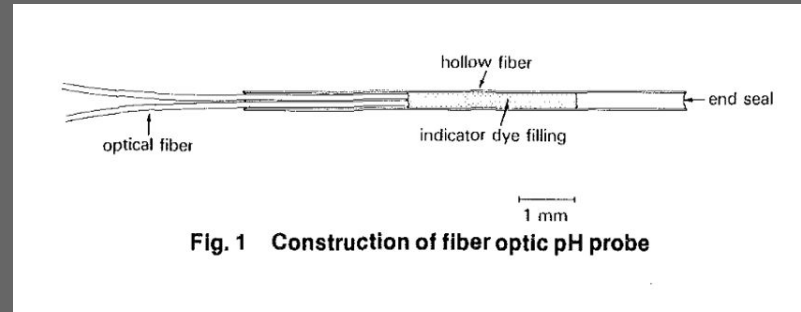


[13]

	Glucose Probe	pH Polymer System	pH ISFET Probe	pH Microsensor	pH Optic Sensor
Size (25)	3/5 (6)	3/5 (15)	2/5 (10)	5/5 (25)	5/5 (25)
Safety (20)	5/5 (20)	5/5 (20)	4/5 (15)	4/5 (15)	5/5 (20)
Accuracy/ Reliability (15)	3/5 (9)	5/5 (15)	4/5 (12)	5/5 (15)	5/5 (15)
Ease of Use (calibration) (15)	1/5 (3)	3/5 (9)	5/5 (15)	5/5 (15)	5/5 (15)
Ease of Fabrication (10)	2/5 (4)	4/5 (8)	2/5 (4)	1/5 (2)	4/5 (8)
Sterility (10)	5/5 (10)	5/5 (10)	5/5 (10)	3/5 (6)	3/5 (6)
Cost (5)	5/5 (5)	3/5 (3)	2/5 (2)	1/5 (1)	3/5 (3)
Efficacy (0)	N/A	N/A	N/A	N/A	N/A
Totals	57	80	68	79	92

Proposed Final Design

- Probe with 2-3 optic fibers
 - Different colors come out of each fiber
- Probe tip will be connected to photodiode or spectrometer to measure wavelength
- Calculate the pH through calculation of the ratios of acidic and basic components of dye.
 - A pH indicator contains a ratio of acidic ions and basic ions
 - These ions reflect different wavelengths of light



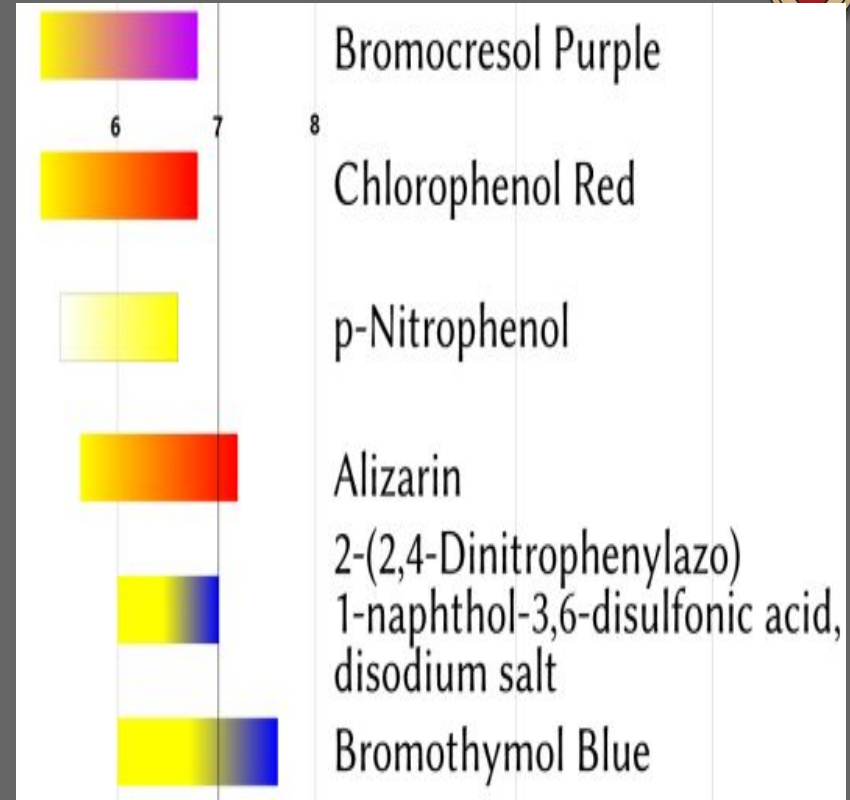


Future Work

- pH indicator

The pH range should be 5-7

- Bromocresol Purple 5.2-6.8
- Bromothymol Blue 6.0-7.6





Immobilization

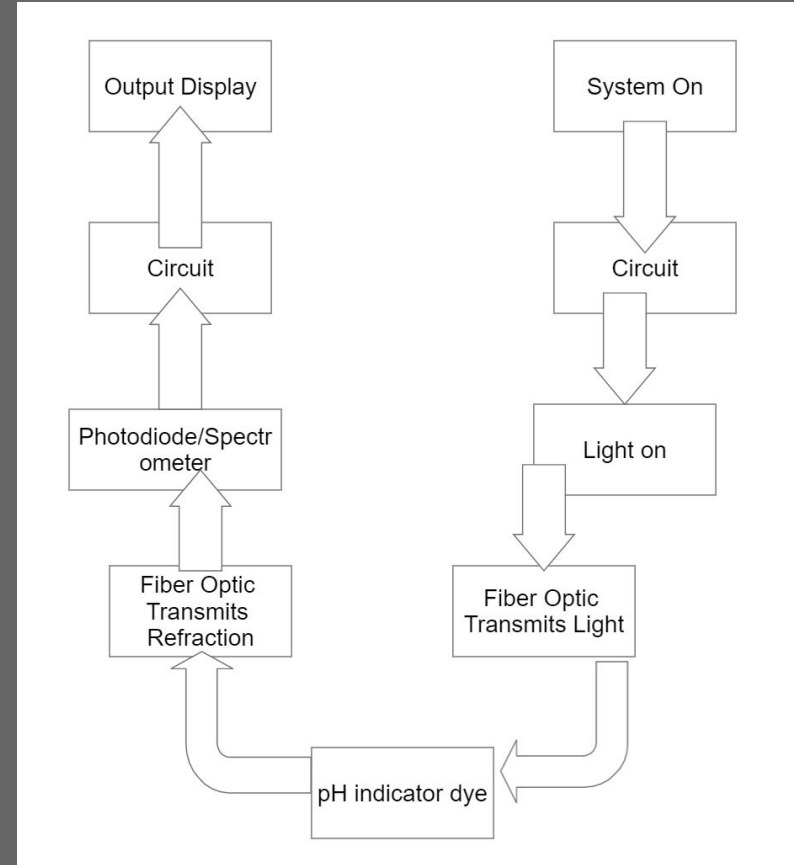
- Connection to the end of the optic fiber
- Sensitive coating on the surface of the sensor
 - Covalent chemical linking
 - Physical encapsulation techniques

- The potential solutions:
 - PTFE
 - Sol-gel fabrication

	Physical	Chemical
Disadvantages	Not effective Leaching problems	Difficulty of fabrication Time consuming
Advantages	Simple and versatile	Excellent immobilization of the pH-sensitive dye No leaching situation

Programming

- Hardware
 - Selection and fabrication of spectroscopy
 - Develop circuit
- Software
 - Coding and data analysis (spectrum and light intensity analysis)
 - Calibration testing (For short calibration time)
 - Sensoring testing (For accuracy)



Acknowledgements



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Questions?

